

SCIENCE Y9- CURRICULUM PLANNING SEQUENCE

ect Year	Term	Big Idea	Торіс	Teacher I Subject Learning Checklist	Trilogy RP	B ig picture	Торіс	Teacher 2 Subject Learning Checklist	Trilogy I
			4.1.1 Cell structure	4.1.1.1 Eukaryotes and prokaryotes			4.7.1 Adaptations, interdependence and	4.7.1.1 Communities	
				4.1.1.2 Animal and plant cells	RPI		competition	4.7.1.2 Abiotic factors	
		≥		4.1.1.3 Cell specialisation				4.7.1.3 Biotic factors	RP7
		Biology		4.1.1.4 Cell differentiation				4.7.1.4 Adaptations	
		0		4.1.1.5 Microscopy		Ecology	4.7.2 Organisation of an ecosystem	4.7.2.1 Levels of organisation	
		<u> </u>		4.1.2.1 Chromosomes		- d	,	4.7.2.2 How materials are cycled	
		e l	4.1.2 Cell division	4.1.2.2 Mitosis and the cell cycle		Ŭ	4.7.3 Biodiversity and the effect of human	4.7.3.1 Biodiversity	
	LS	မီ		4.1.2.3 Stem cells		→ E	interaction on ecosystems	4.7.3.2 Waste management	
	-28hrs	Ŭ	4.1.3 Transport in cells	4.1.3.1 Diffusion	1 1		interaction on ecosystems	4.7.3.3 Land use	
	5	4 		4.1.3.2 Osmosis	RP2			4.7.3.4 Deforestation	
	<u> </u>	V		4.1.3.3 Active transport				4.7.3.5 Global warming	
	3							4.7.3.6 Maintaining biodiversity	
	erm		4.2.1 Principles of organisation	4.2.1 Principles of organisation	+	5	4.2.2 Animal tissues, organs and organ	4.2.2.2 The heart and blood vessels	
	ΗĔ	ç	· ·			rganisatior		4.2.2.3 Blood	
		satio	4.2.2 Animal tissues, organs and	4.2.2.1 The human digestive system		lisa	systems		
		gani	organ systems			gai		4.2.2.4 Coronary heart disease: a non-communicable disease	
		Ď			RP3 & 4	4.2 Or		4.2.2.5 Health issues	
		4.2						4.2.2.6 The effect of lifestyle on some non-communicable diseases	
								4.2.2.7 Cancer	
		<u>+</u> -	4.1.1 A simple model of the atom,	4.1.1.1 Atoms, elements and compounds			4.1.1 Energy changes in a systme, and the	4.1.1.1 Energy stores and systems	
2		4	symbols, relative atomic mass,	4.1.1.2 Mixtures			ways energy is stored before and after these		
	11	is cal	4.8.1 Purity, formulations and	4.8.1.1 Pure substances		SS .	changes	4.1.1.3 Energy changes in systems	1
:		4.8 Chemic analysi	chromatography	4.8.1.2 Formulations		Energy		4.I.I.4 Power	
;				4.8.1.3 Chromatography	RP12		4.1.2 Conservation and dissipation of energy	4.1.2.1 Energy transfers in a system	
	11	U	4.1.1 A simple model of the atom,	4.1.1.3 The development of the model of the atom	1 1	4	,	4.1.2.2 Efficiency	
cear 9		the periodic	symbols, relative atomic mass,				4.1.3 National and global energy resources	4.1.3 National and global energy resources	
ear			electronic charge and isotopes	4.1.1.4 Relative electrical charges of subatomic particles					
, ∥⊁				4.1.1.5 Size and mass of atoms		model of	4.3.1 Changes of state and the particle	4.3.1.1 Density of materials	
		ld ti		4.1.1.6 Relative atomic mass			8	4.3.1.2 Changes of state	
	LS	ear		4.1.1.7 Electronic structure			model 4.3.2 Internal energy and energy transfers	4.3.2.1 Internal energy	
	8	ucture table		4.1.2.1 The periodic table			4.5.2 Internal energy and energy transfers	4.3.2.2 Temperature changes in a system and specific heat capacity	
5	-28hr		4.1.2 The periodic table	4.1.2.2 Development of the periodic table		atte		4.3.2.3 Changes of heat and specific latent heat	
)	5	c Str		4.1.2.3 Metals and non-metals		Particle	4.3.3 Particle model and pressure	4.3.3.1 Particle motion in gases	
	3	Atomic		4.1.2.4 Group 0		Pa	1.5.5 Farticle moder and pressure		
	erm			4.1.2.5 Group 1		.4 .0			
	ΗĔ	4.		4.1.2.6 Group 7		4		4.4.1.1 The structure of an atom	
				4.2.1.1 Chemical bonds			4.4.1 Atoms and isotopes	4.4.1.2 Mass number, atomic number and isotopes	
		ructure rties of	4.2.1 Chemical bonds, ionic, covalent and metallic	4.2.1.2 Ionic bonding		e ji		4.4.1.3 The development of the mode of the atom	
				4.2.1.3 Ionic compounds		Atomic ucture	4.4.2 Atoms and nuclear radiation	4.4.2.1 Radioactive decay and nuclear radiation	
		, str oper ter		4.2.1.4 Covalent bonding				4.4.2.2 Nuclear equations	
		iding, e prop matte		4.2.1.5 Metallic bonding		4.4 str		4.4.2.3 Half lives and the random nature of radioactive decay	
		Bone I the	122 Herrichanding and atmost	4.2.2.1 The three states of matter		1		4.4.2.4 Radioactive contamination	
		4.2 E and	4.2.2 How bonding and structure						
			are related to the properties of	4.2.2.2 State symbols			4.2.1 Current, potential difference and	4.2.1.1 Standard circuit diagram symbols	
	s	anic try	4.7.1 Carbon compounds as fuels and feedstock	4.7.1.1 Crude oil, hydrocarbons and alkanes			resistance	4.2.1.2 Electrical charge and current	
	²	Orgar emistr		4.7.1.2 Fractional distillation and petrochemicals		ity		4.2.1.3 Current, resistance and potental difference	
	24hrs	4.7 C cher		4.7.1.3 Properties of hydrocarbons 4.7.1.4 Cracking and alkenes		rici		4.2.1.4 Resistors	
		4 ~		4.7.1.4 Cracking and alkenes 4.10.1.1 Using the Earth's resources and sustainable development	┼───┤	Electricity	4.2.2 Series and parallel circuits	4.2.2 Series and parallel circuits 4.2.3.1 Direct and alternating potential differences	
	m	s a	4.10.1 Using the Earth's resources				4.2.3 Domestic uses and safety		
	3	Usin urce	and obtaining potable water	4.10.1.2 Potable water	RP13	4.2		4.2.3.2 Mains electricity	
	Term	4.10 L resou		4.10.1.3 Waster water treatment			4.2.4 Energy transfers	4.2.4.1 Power	
	ΗĔ	-4 -	4.10.2 Life cycle assessment and	4.10.2.1 Life cycle assessment				4.2.4.2 Energy transfers in everyday appliances	
			recycling	4.10.2.2 Ways of reducing the use of resources				4.2.4.3 The National Grid	
		— .	I		1	F			
		Biology	Required practical activity 1:	Use a light microscope to observe, draw and label a selection of plant and animal cells. A magnificat scale must be included.		Biology	Required practical activity 14:	Measure the population size of a common species in a habitat. Use sampling techniques to	o Investigat
								effect of a factor on the distribution of this species.	
			Required practical activity 2:	Investigate the effect of a range of concentrations of salt or sugar solutions on the mass of plant		Dhunic -		A in the interview of the second s	
						Physics		An investigation to determine the specific heat capacity of one or more materials. The inv	-
				tissue.			1	involve linking the decrease of one energy store (or work done) to the increase in tempe	erature and
			Required practical activity 3:	Use qualitative reagents to test for a range of carbohydrates, lipids and proteins.			Description of the second s	subsequent increase in thermal energy stored.	X
							Required practical activity 15:	Use circuit diagrams to set up and check appropriate circuits to investigate the factors aff	-
								resistance of electrical circuits. This should include: • the length of a wire at constant temperature	
								combinations of resistors in series and parallel.	
		Required practical activity 4: . Investigate the effect of pH on the rate of reaction of amylase enzyme		7		Required practical activity 16:	Use circuit diagrams to construct appropriate circuits to investigate the I–V characteristic	cs of a varie	
							circuit elements, including a filament lamp, a diode and a resistor at constant temperature	e.	
		Chemistry	Required practical activity 12:	Investigate how paper chromatography can be used to separate and tell the difference	between		Required practical activity 17:	Use appropriate apparatus to make and record the measurements needed to determine t	
				coloured substances. Students should calculate Rf values.				regular and irregular solid objects and liquids. Volume should be determined from the dimensions of	
			Required practical activity 13:	Analysis and purification of water samples from different sources, including pH, dissolve	ed solids and				
		1		distillation.					